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Chemistry 1001: Solutions: A Special Type of Mixture ...

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Solutions: A Special Type of Mixture: Students learn to describe and explain the process of forming a solution using the terms solute and solvent and to compare the properties of suspensions, colloids, and solutions. The factors that affect the rate at which a solid solute dissolves in a liquid solvent are introduced.

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4Cl are dissolved at 68 °C, the solution is _____. 3. If 54 g of NH₄Cl are dissolved at 30 °C, how many grams don't dissolve? _____ Answer each of the following questions about molarity. Show all work on the problems. 1. Describe, IN DETAIL, how to make one liter of a 1 M NaCl solution. 1 s 2. What is the molarity of a solution that contains ...

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3. Solutions are said to be in a _____ phase even though the components may have been in different phases before the solution was formed. 4. Pure gold is 24 carat. 14-carat gold contains 14 parts gold and 10 parts other metals. 14-carat gold is said to be a(n) _____, which is a type of _____ solution. An example of a gaseous solution is

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